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⑤④ **Crystalline human granulocyte colony stimulating factor and process for preparing the same.**

⑤⑦ Crystalline human granulocyte colony stimulating factor (crystalline G-CSF) and a process for preparation of the same are disclosed. Crystalline G-CSF is extremely pure and exhibits high activity and low antigenicity. In addition, the crystallization of G-CSF improves its stability and makes handling thereof easy when storing and formulating it into desired dosage forms of drugs.

EP 0 344 796 A2

CRYSTALLINE HUMAN GRANULOCYTE COLONY STIMULATING FACTOR AND PROCESS FOR PREPARING THE SAME

Human granulocyte colony stimulating factor (hereinafter referred to as human G-CSF) acts upon human bone marrow cells and induces their differentiation and proliferation into granulocytes. Human G-CSF is expected to be applied as a drug for chemotherapy or prevention of various kinds of leukemia and as a therapeutic for serious infectious diseases which cannot be treated with antibiotics alone.

5 Several types of human G-CSF have been reported. Natural human G-CSF was isolated from the culture supernatant of G-CSF-producing cells which were established from human tumor cells in patients with oral cancer (Japanese Patent Public Disclosure No. 62-227526). G-CSF free from sugar chain was produced by *E. coli* used as host cells by means of recombinant DNA technology (Japanese Patent Public Disclosure No. 62-132898 and Japanese Patent Public Disclosure No. 62-132899). Glycoprotein G-CSF was

10 also produced by using animal cells as a host (Japanese Patent Public Disclosure No. 62-236497). Many kinds of techniques have been developed for the crystallization of proteins or polypeptides. In particular, crystallization of enzymes is in fact carried out as being one of the most effective procedures for raising the purity. However, no generalized procedure for crystallization has been developed. The species of enzymes, the conditions of crystallization such as pH, ionic strength, temperature and protein concentration

15 may vary. Furthermore, some bioactive proteins or polypeptides are very unstable in the face of changes in pH, ionic strength, temperature and so on, which leads to potential changes in three dimensional structure and results in a reduction or total loss of activity. It is therefore necessary to choose appropriate conditions for crystallization which will not lead to any change in three dimensional structure.

20 It is commonly known that many factors which have to be determined by a great deal of trials and errors are involved in protein crystallization, and thus, in many cases of crystallization, a lot of trials are necessary.

In order to crystallize proteins from an aqueous solution, it is necessary to gradually increase the stronger affinity between protein molecules than between the protein and water. With respect to molecules like proteins having a high molecular weight and having surfaces with complex physical properties, it is of course natural that the law that governs minimization of the free energy of the interaction between proteins holds, but the interaction between proteins is much more complex than that between organic compounds having low molecular weights. It is considered that the various factors which cause interactions between proteins include electronic charges, steric effect, hydrophobic effect, hydrophilic effect and Van-der-Waals effect. The interactions between proteins, and between proteins and water function in such a way that a system containing those proteins is placed in the most stable state. Thus, protein molecules show various types of mutual interaction depending on the concentration, temperature, pH, ionic strength and so on, and in some cases proteins can crystallize while in other cases they cannot crystallize, resulting only in aggregation. Even in a case where protein crystallizes, polymorphism of crystals may be produced

35 depending on such factors as protein concentration, temperature, ionic strength and so on. It is known that crystallization of proteins of low purity is very difficult to achieve.

As described above, many trials for the crystallization of proteins and polypeptides have been carried out and the crystallization of some proteins has, in fact, been successful. However, the most of successful cases were limited to just proteins among the great number of naturally occurring proteins and were also

40 limited to very stable proteins such as exocellular proteins and proteins isolated from plants. The following important advantages are obtained by crystallizing human G-CSF, which has previously thought to have been very difficult. These advantages are:

- (1) to heighten the purity and to reduce the antigenicity;
- (2) to increase the stability thereby making handling of G-CSF easy during storage and drug
- 45 preparation; and
- (3) to increase the activity of G-CSF.

Fig. 1 is a microscopic photograph (x 100) of crystalline human G-CSF prepared by the method described in Example 1.

50 Fig. 2 shows the colony formation stimulating activity of crystalline human G-CSF prepared by the method described in Example 1.

Figs. 3a and 3b show HPLC chromatographs of crude human G-CSF and crystalline human G-CSF, respectively [detection, absorbance at 200 nm; elution buffer, 10 mM phosphate buffer (pH 6.8) + 100 mM NaCl; column, GPC (TSK G-3000); flow rate, 1 ml/min.]

Fig. 4 is a precession photograph of (OK1) crystals I and II ($\mu = 10^\circ$) of crystalline human G-CSF prepared by the method of Example 1.

Fig. 5 shows a microscopic photograph (x 84) of crystalline human G-CSF prepared by the method described in Example 5.

Many types of human G-CSF purified by the methods described above (Japanese Patent Public Disclosure Nos. 61-227526, 62-132898, 62-132899 and 62-236497) are dissolved in a neutral or weak acidic buffer and crystallized in the presence of a crystallizing agent, and crystals are then recovered by centrifugation.

It is preferable for crystalline human G-CSF to be purified to such an extent that it contains neither any impurities nor any polymers like dimers or trimers when dissolved in water.

Human G-CSFs which may be employed as starting materials in the instant crystallization process include a natural human G-CSF which is a glycoprotein, a human G-CSF which is produced by animal cells as a host by recombinant DNA technique and its derivatives, and a human G-CSF free from sugar chains which is produced by *E. coli* as a host by recombinant DNA technique and its derivatives. Generally, crystallization of glycoproteins and proteins having sugar chains is more difficult than that of proteins having no sugar chains, but the instant crystallization procedure makes it possible to crystallize glycoprotein human G-CSFs.

The crystallization of human G-CSF by the instant invention can be carried out by gas-vapor equilibrium methods such as the hanging drop method and the depression slide method; methods using precipitants such as the button method, the dialyzing membrane method and the batch method; and dialysis methods such as the button method and the dialyzing membrane method.

Each of these methods is illustrated below.

Gas-vapor equilibrium method

Representative examples of this method are the hanging drop method and the depression slide method. In both methods, a human G-CSF to be crystallized is dissolved in weak acidic or neutral buffers preferably having a pH of from 3.0 to 7.0 such as acetate buffer, phosphate buffer, citrate buffer, phosphate-citrate buffer with the concentration of from 5 to 100 mM. It is preferred that the concentration of human G-CSF in a buffer is in the range of from about 1 to 100 mg/ml.

A crystallizing agent such as neutral or weak acidic salts e.g. $(\text{NH}_4)_2\text{SO}_4$, Na_2SO_4 , NaCl , KCl , NH_4Cl , NaH_2PO_4 , KH_2PO_4 , $\text{NH}_4\text{H}_2\text{PO}_4$, LiCl , CsCl , CaCl_2 , MgCl_2 , MgSO_4 and so on and, then, an organic compound e.g. polyethyleneglycol are added to the buffer containing human G-CSF as described above. Although the concentration of the precipitating agent varies depending on the kind of agent and buffer, $(\text{NH}_4)_2\text{SO}_4$ is used in a concentration of about 1 to 20%, preferably about 2 to 15% of the saturation level. As for other precipitating agents, a concentration of about 0.1 to 1.5 M, preferably about 0.2 to 1.2M is used for crystallization.

Furthermore, if necessary, a stabilizing agent (e.g. neutral salts such as NaCl , KCl , CsCl , MgCl_2 and CaCl_2 or organic compounds such as glycerol, dioxan and acetone), an antiseptic (e.g. NaN_3), a chelating agent for heavy metals (e.g. EDTA), a thiol group protecting agent or an antioxidant (e.g. mercaptoethanol, cysteine, glutathione and dithiothreitol), an antipolymerizing agent, surfactant (e.g. octyl- β -glycoside, hexantriol, tweens or octaethyleneglycol-n-dodecylether) and so on can be added to the protein solution.

In the hanging drop method, droplets of human G-CSF solution prepared by the method described above are put on the back face of a cover glass plate, the glass plate is set on a reservoir which contains a predetermined amount of precipitant and a buffer, usually, the same as that used in preparation of the human G-CSF solution, and the reservoir is sealed completely with silicone grease.

In the depression slide method, a predetermined amount of a buffer and precipitant is placed in a reservoir, human G-CSF solution is put on a depression slide glass and the reservoir is allowed to stand at 5 to 30°C for 1 to 21 days preferably in an quake-proof incubator and crystalline human G-CSF is thus produced.

Salting-out method using dial (Precipitant method)

One to 100 mg/ml human G-CSF solution in a buffer is put in a dialysis tube and is dialyzed against the same buffer containing 10 to 14% saturated precipitating agent [e.g. 10 to 14% saturation of $(\text{NH}_4)_2\text{SO}_4$, 0.2

to 1.5M NaCl, 0.2 to 1.5M NaH₂PO₄ and so on] or polyethyleneglycol solution until crystal grow (usually for 1 week). After that the crystals produced can be isolated by centrifuging the contents in the dialysis tube.

5 Button method and Dialysis method using dialysis membrane

One to 100 mg/ml human G-CSF in a buffer solution is put in a button for crystallization, is sealed with a EDTA-treated dialysis tube and is dialyzed against the same buffer of low concentration for 1 day to 1 week. After that the needle-shaped crystals produced can be isolated by centrifugation of the contents of
10 the button.

Batch method

15 One to 100 mg/ml human G-CSF in a buffer solution is put in suitable containers, and a solution of a precipitating agent which is usually dissolved in the same buffer is added to the containers. The resulting solution is stirred as necessary, and the mixture is allowed to stand for about 1 to 2 weeks to produce crystalline human G-CSF as colorless plate crystals.

The crystals can be isolated by centrifugation from the buffer solution in which they are contained.
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Characterization of crystals

The crystalline human G-CSF obtained by this invention was analyzed by X-ray diffraction and three
25 kinds of crystals having the following characteristics were identified.

30	I	Crystal form:	orthorhombic			
		Space group:	P2 ₂ 1 ₂ 1			
		Length of axes (Å):	a	29, b	94, c	111
		Solvent volume (%):	Vs	37		
35	II	Crystal form:	orthorhombic			
		Space group:	P2 ₁ 2 ₁ 2 ₁			
		Length of axes (Å):	a	28, b	100, c	110
		Solvent volume (%):	Vs	37		
40	III	Crystal form:	orthorhombic			
		Space group:	P2 ₂ 1 ₂ 1			
		Length of axes (Å):	a	29, b	107, c	114
		Solvent volume (%):	Vs	46		

Precession photograph of (OKL) of crystals I and II ($\mu - 10^\circ$) is given as Fig. 4.

In some cases chimeracrystals of the two crystalline substances I and II described above are produced.

45 Measurement of activity of crystalline human G-CSF

In order to measure the activity of the crystalline human G-CSF, colony stimulating activity (CSA) was assayed in vitro by the following method.

50 (Assay method)

Assay for CSA was carried out by the soft agar mono-layer cultivation method as in Bradley T. R. and Metcalf D. et al. (Aust. J. Exp. Biol. Med. Sci., 44, 287 - 300, 1966). That is, a mixture of 1 ml of horse serum (Hyclone Co., Ltd.), 0.25 ml of test sample, 0.25 ml of C57BL mouse myeloid cell suspension (1.5×10^6 cells/ml) and 1 ml of modified McCoy's 5A medium containing 0.75% agar was poured into a plastic dish for tissue culture. After gelation, the dish was incubated at 37°C or 5°C at 100% R.H. in an atmosphere of 5% CO₂/95% air for 5 days, and the number of colonies formed was counted (one colony
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comprises more than 50 cells). One unit of CSA was defined as being the activity required to form one colony.

A test sample (original solution) was prepared as follows: 10 μ l of a crystalline human G-CSF solution (14 mg/ml, quantitatively analyzed by HPLC) in 100 mM acetate buffer (pH 4.2) was diluted with phosphate buffered saline containing 0.5% mouse serum albumin (Cappel Co., Ltd.) (0.5% mSAPBS) to adjust the concentration to 100 μ g/ml (total volume was 1.4 ml). After filtration using a Millipore filter, respective samples of this original solution are diluted 10, 100, 500, 2,500, 12,500 and 62,500 times, respectively, with 0.5% mSAPBS. Also 0.5% mSAPBS was used as a comparison and 1 μ g/ml of rG-CSF (noncrystalline, purified human G-CSF produced by recombinant DNA technique) was prepared for comparative purposes.

The results are as follows (also shown in Fig. 2):

Dilution	The number of colonies formed		
	Experiment 1	Experiment 2	Average
1	51	55	54
10	46	47	46.5
100	44	X	44
500	30	59	44.5
2,500	18	27	22.5
12,500	6	7	6.5
62,500	5	3	4
rG-CSF	29	44	36.5
Control	5	7	6

As clearly shown from the above data, the crystalline human G-CSF of this invention exhibited the same or higher colony stimulating activity than that of noncrystalline and human G-CSF produced by recombinant DNA technique.

A detailed explanation of this invention is set forth below in regard to Examples. By a similar method, another human G-CSF which exhibits a high degree of sequence homology can be crystallized.

Example 1

Depression slide method (gas-vapor equilibrium method)

A solution of glycoprotein human G-CSF (10 μ l, 15.5 mg/ml), which had been produced by animal cells as a host by recombinant DNA technique, was dissolved in 10 mM citrate buffer (pH 4.6) and the solution was mixed with 16% saturation of a $(\text{NH}_4)_2\text{SO}_4$ solution (10 μ l) on a depression slide glass. One ml of 12% saturation of $(\text{NH}_4)_2\text{SO}_4$ solution in 10 mM citrate buffer (pH 4.6) was put in a reservoir and the reservoir was covered with a glass plate and sealed with silicone. After the reservoir had been allowed to stand for about 3 weeks at room temperature (20 to 24°C), single crystals of human G-CSF in the form of a colorless, transparent plate were obtained (Fig. 1).

In order to determine the characteristics of the crystals, the signal crystals so produced and the mother liquid were put and sealed in a quartz capillary tube and were analyzed by X-ray diffraction.

The results of the analyses allowed three kinds of crystals having the following characteristics to be identified.

I	Crystal form:	orthorhombic			
	Space group:	P2 ₁ 2 ₁			
	Dimension of unit crystal lattice (Å):	a	29, b	94, c	111
	Number of molecules per unit crystal lattice:	8			
	Volume of unit crystal lattice (Å ³):	303,900			
	Solvent volume (%):	Vs	37		
II	Crystal form:	orthorhombic			
	Space group:	P2 ₁ 2 ₁ 2 ₁			
	Dimension of unit crystal lattice (Å):	a	28, b	100, c	110
	Number of molecules per unit crystal lattice:	8			
	Volume of unit crystal lattice (Å ³):	303,600			
	Solvent volume (%):	Vs	37		
III	Crystal form:	orthorhombic			
	Space group:	P2 ₁ 2 ₁			
	Dimension of unit crystal lattice (Å):	a	29, b	107, c	114
	Number of molecules per unit crystal lattice:	8			
	Volume of unit crystal lattice (Å ³):	353,742			
	Solvent volume (%):	Vs	46		

The produced crystal were washed twice with the buffer containing 25% saturation of (NH₄)₂SO₄ and then analyzed by HPLC [column, GPC (TSK G-3000); elution buffer, 10 mM phosphate buffer (pH 6.8) + 100 mM NaCl; flow rate, 1 ml/min.] with a retention time of 19.75 minutes. The results of HPLC analyses are shown in Fig. 3a (control) and 3b, which demonstrate that these crystals were G-CSF.

Example 2

Hanging drop method (gas-vapor equilibrium method)

Solutions of the glycoprotein human G-CSF (2 µl, 18.3 mg/ml) used in Example 1 dissolved in 100 mM acetate buffer (pH 4.2) in which a precipitating agent and/or a stabilizing agent had been dissolved were mixed with various concentrations of NaH₂PO₄ solution (2 µl, 0.8, 0.9, 1.0, 1.1 and 1.2 M) on silanized cover glasses. Reservoirs each containing 8 ml of various concentrations of NaH₂PO₄ solutions (0.7, 0.8, 0.9, 1.0, 1.1 and 1.2 M) were covered with the cover glasses, sealed with silicone grease and allowed to stand. Among the various combinations of these samples, colorless plate crystals separated out in about 24 hours in the earliest case.

The crystal which separated out in about 2 weeks was identified as G-CSF by means of HPLC.

The conditions for crystallization of each sample are summarized in the table below.

Buffer	Crystallizing agent	Reservoir	Stabilizing agent and others	
100 mM acetate buffer (pH 4.2)	5 - 15% saturated (sat.) (NH ₄) ₂ SO ₄ (AS)	10 - 20% sat. AS	none	
the same as above	8 - 9% sat. AS 0.4 - 0.8% CsCl	10 - 20% sat. AS	none	
the same as above	5 - 15% sat. AS	10 - 20% sat. AS (5% glycerol)	glycerol	
the same as above	0.3 - 0.5 M NaCl	0.5 - 1.0 M NaCl	none	
10 mM citrate buffer (pH 4.2)	8 - 9% sat. AS	10 - 15% sat. AS	none	
10 mM citrate/phosphate buffer (pH 5.4)	8.0 - 12.0% sat. AS	10 - 18% sat. AS	none	
10 mM citrate/phosphate buffer (pH 6.4)	8.0 - 11.5% sat. AS	10 - 14% sat. AS	none	
100 mM citrate buffer (pH 3.0)	6.0 - 10% sat. As	10 - 14% sat. As	none	
100 mM phosphate buffer (pH 6.8)	0.6 - 0.8 M NaH ₂ PO ₄	0.7 - 1.0 M NaH ₂ PO ₄	100 mM NaCl	
100 mM acetate buffer (pH 4.2)	8.5 - 12% polyethyleneglycol 4000	10 - 15% polyethyleneglycol 4000	100 mM	MgCl ₂
			10 mM	CaCl ₂
			0.01%	NaN ₃
			1 mM	EDTA
the same as above	8.5 - 12% polyethyleneglycol 6000	10 - 15% polyethyleneglycol 6000	100 mM	MgCl ₂
			0.5 mM	CaCl ₂
			0.01%	NaN ₃

Example 3Dialysis membrane method

A solution of human G-CSF as used in Example 1 dissolved in 100 mM acetate buffer (pH 4.2) (1050 μ l, 16.3 mg/ml) was sealed in a dialysis tube (pore size: MW 7,000 cut-off, EDTA-pretreated). This tube was soaked in 200 ml of 1 M KCl containing 10 mM acetate buffer (pH 4.2) and was subjected to salting-out for about 1 week with outer solvent being stirred gently. The contents were then transferred to a V-shaped vial and colorless plate crystals were recovered by centrifugation (5° C, 15,000 rpm, 10 minutes).

Example 4

Batch method

(1) Human G-CSF as used in Example 1 dissolved in 100 mM acetate buffer (pH 4.2) (200 μ l, 15.58 mg/ml) was put in a container. To this solution was added a 100 mM acetate buffer containing 1.2M NaH_2PO_4 and the mixture was allowed to stand for 2 weeks at room temperature.

Colorless plate crystals separated out and were then isolated by centrifugation (5 °C, 15,000 rpm, 10 minutes).

(2) As described in (1), a human G-CSF solution in 100 mM acetate buffer (pH 4.2) (300 μ l, 13.08 mg/ml) was mixed with 300 μ l of a 100 mM acetate buffer containing 16% saturation of $(\text{NH}_4)_2\text{SO}_4$ and the mixture was allowed to stand for 3 weeks. Colorless plate crystals were isolated by centrifugation.

Example 5

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Dialysis membrane method

Human G-CSF as used in Example 1 which had been dialyzed against a 15% saturation of $(\text{NH}_4)_2\text{SO}_4$ solution was dissolved in 5 mM citrate buffer (pH 4.6). The human G-CSF solution (200 μ l) was put in a dialysis tube and was dialyzed against 80 ml of water. As shown in Fig. 4, very long needle crystals were produced.

Example 6

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Hanging drop method (gas-vapor equilibrium method)

Human G-CSF as used in Example 1 which had been dialyzed against a 15% saturation of $(\text{NH}_4)_2\text{SO}_4$ was dissolved in a 5 mM citrate buffer containing 3 mM octyl- β -glycoside (pH 4.6). Five ml of this human G-CSF solution was mixed with 5 ml of a 15 to 22% saturation of $(\text{NH}_4)_2\text{SO}_4$ solution or with 5 ml of a 5 mM sodium citrate solution. The human G-CSF in the solution was crystallized with use of 8 ml of a 11 to 13% saturation of $(\text{NH}_4)_2\text{SO}_4$ or a 5 mM sodium citrate solution as an outer solution. Plate crystals were produced in several days to 2 weeks.

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Example 7Dialysis membrane method

A solution of human G-CSF as used in Example 1 in a 10 mM acetate buffer (pH 4.2) (610 μ l, 11.45 mg/ml) was placed in dialysis tube and then dialyzed against 200 ml of 1 M KCl solution containing a 10 mM acetate buffer (pH 4.2). The KCl concentration of the outer solution was increased from the next day at a rate of 0.1 M per day and finally reached 1.4 M. The contents in dialysis tube were then transferred to a V-shaped vial and centrifuged (10 °C, 15,000 rpm, 5 minutes) to remove the liquid phase. Colorless plate crystals were isolated. These crystals were dissolved in 400 μ l of a 10 mM acetate buffer (pH 4.2) and centrifuged (10 °C, 12,000 rpm, 5 minutes). The concentration of G-CSF in the resulting supernatant (420 μ l) was determined to be 15.05 mg/ml (yield 90.7%) by HPLC analysis.

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Example 8Dialysis membrane method

A solution of human G-CSF as used in Example 1 in a 10 mM acetate buffer (pH 4.2) (680 μ l, 32.09 mg/ml) was poured in dialysis tube and then dialyzed against a 10 mM acetate buffer containing 5%

saturation of $(\text{NH}_4)_2\text{SO}_4$ (200 μl). The $(\text{NH}_4)_2\text{SO}_4$ concentration of outer solution was gradually increased from the next day at a rate of 1% per day and finally reached 13%. The contents of the tube were transferred to a V-shaped vial and then centrifuged (10 °C, 15,000 rpm, 5 minutes) to remove the liquid phase whereby colorless plate crystals were obtained. The crystals were dissolved in 1,200 μl of a 10 mM acetate buffer (pH 4.2) and the solution was centrifuged (10 °C, 12,000 rpm, 5 minutes). The concentration of G-CSF in the resulting supernatant (1290 μl) was determined to be 16.32 mg/ml (yield 96.4%) by HPLC analysis.

10 Example 9

Batch method

15 To a solution of human G-CSF as used in Example 1 in a 10 mM acetate buffer (pH 4.2) (200 μl , 19.84 mg/ml) was added a 1.5 M NaH_2PO_4 solution (250 μl) and the mixture was allowed to stand for 1 week. The contents were then transferred to a V-shaped vial and colorless plate crystals were isolated by centrifugation (10 °C, 15,000 rpm, 10 minutes). The collected crystals were dissolved in 400 μl of a 10 mM acetate buffer and the solution was centrifuged (10 °C, 12,000 rpm, 10 minutes). The concentration of G-CSF in the resulting supernatant (450 μl) was determined to be 12.51 mg/ml (yield 78.8%) by HPLC analysis.

25 Example 10

Batch method

30 To a solution of human G-CSF as used in Example 1 in a 10 mM acetate buffer (pH 4.2) (1,200 μl , 28.91 mg/ml) on ice was added a 34% saturation of $(\text{NH}_4)_2\text{SO}_4$ solution (800 μl). The mixture was allowed to stand at 14 °C and the temperature was gradually increased to 24 °C at a rate of 1 °C per day. The contents were transferred to a V-shaped vial and colorless plate crystals were isolated by centrifugation (10 °C, 15,000 rpm, 5 minutes) to remove the liquid phase. The crystals were dissolved in a 10 mM acetate buffer (pH 4.2) and the solution was centrifuged (10 °C, 12,000 rpm, 5 minutes). The concentration of G-CSF in the resulting supernatant (1380 μl) was determined by HPLC and found to be 24.05 mg/ml (yield 95.7%).

Crystalline human G-CSF prepared by the instant invention was successfully obtained in a high yield (78.8 to 96.4%) as shown in Example 7 to 10. Furthermore, as shown in Fig. 2, crystalline human G-CSF prepared according to this invention exhibited colony stimulating activity that was the same as or higher than that of noncrystalline human G-CSF. From the view point of this activity, this invention is very useful for applications in the medical and pharmaceutical fields. The human G-CSF of this invention also offers great advantages in allowing of easy formulation into dosage forms because it is in a crystalline form and is very stable.

45 Example 11

Button method

50 Human G-CSF obtained by the salting-out method was dissolved in a 5 mM citrate buffer (pH 4.6) to give a final concentration of about 10 mg/ml. The solution was put in a 30 μl capacity button for crystallization and sealed with EDTA-treated dialysis membrane, carefully avoiding the inclusion of bubbles in it. The membrane was fixed firmly with a rubber ring. This button was dialyzed against a 2 mM citrate buffer to give needle crystals in 1 - 3 days.

Under the following conditions, crystalline human G-CSF was obtained by the same method.

	Buffer		Detergent	Crystal shape
	pH	Concentration (solvent → outer solution)		
5	Hepes buffer (pH 5.6)	20 mM → 3 mM	-	micro
10	same as above	20 mM → 3 mM	1 mM	plate
		20 mM → 5 mM	octyl- β -glycoside	
15	citrate buffer (pH 4.6)	10 mM → 3 mM	-	needle
		10 mM → 5 mM		
20	same as above	5 mM → 3 mM	-	needle
		5 mM → 2 mM		
		5 mM → 1 mM		
25	same as above	10 mM → 5 mM	1 mM	needle
		10 mM → 3 mM	octa-ethyleneglycol-- n-dodesylether	
30	same as above	10 mM → 5 mM	1 mM	needle
		10 mM → 3 mM	octyl- β -glycoside	
35	same as above	5 mM → 3 mM	same as above	needle
		5 mM → 2 mM		
		5 mM → 1 mM		

Claims

1. Human granulocyte colony stimulating factor in crystalline form.
2. Human granulocyte colony stimulating factor in crystalline form according to Claim 1 wherein the human granulocyte colony stimulating factor is produced by animal cells or other eukaryotic cells by recombinant DNA technique.
3. Human granulocyte colony stimulating factor in crystalline form according to Claim 1 wherein the human granulocyte colony stimulating factor is isolated from culture media of human cells which have the ability to produce the granulocyte colony stimulating factor.
4. Human granulocyte colony stimulating factor in crystalline form according to Claim 1 wherein the human granulocyte colony stimulating factor is produced by a prokaryotic microorganisms as a host by recombinant DNA technique.
5. A process for preparing crystalline human granulocyte colony stimulating factor which comprises dissolving a human granulocyte colony stimulating factor and optionally a precipitating agent in a buffer containing 0 to 100 mM of a buffering salt and having a pH of from 3.0 to 7.0, and crystallizing the granulocyte colony stimulating factor by the gas-vapor equilibrium method, dialysis method, button method or batch method.
6. A process for preparing human granulocyte colony stimulating factor in crystalline form according to Claim 5 wherein human granulocyte colony stimulating factor is dissolved in a buffer containing 1 and 100 mg/ml of the buffering salt.
7. A process for preparing human granulocyte colony stimulating factor in crystalline form according to Claim 5 wherein the precipitating agent, when used, is selected from the group consisting of $(\text{NH}_4)_2\text{SO}_4$, Na_2SO_4 , NaCl , KCl , NH_4Cl , NaH_2PO_4 , KH_2PO_4 , $\text{NH}_4\text{H}_2\text{PO}_4$, LiCl , CsCl , CaCl_2 , MgCl_2 , MgSO_4 , polyethyleneglycol, sodium citrate, sodium oxalate and mixtures thereof.

8. A process for preparing a human granulocyte colony stimulating factor in crystalline form according to Claim 7 wherein the concentration of the precipitating agent in the solution containing the human granulocyte colony stimulating factor ranges from 0.1 to 1.5 M.

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Fig. 1

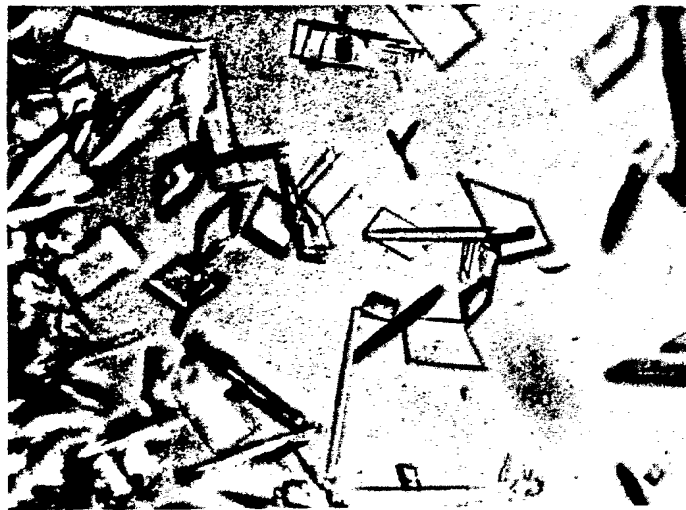


Fig. 2

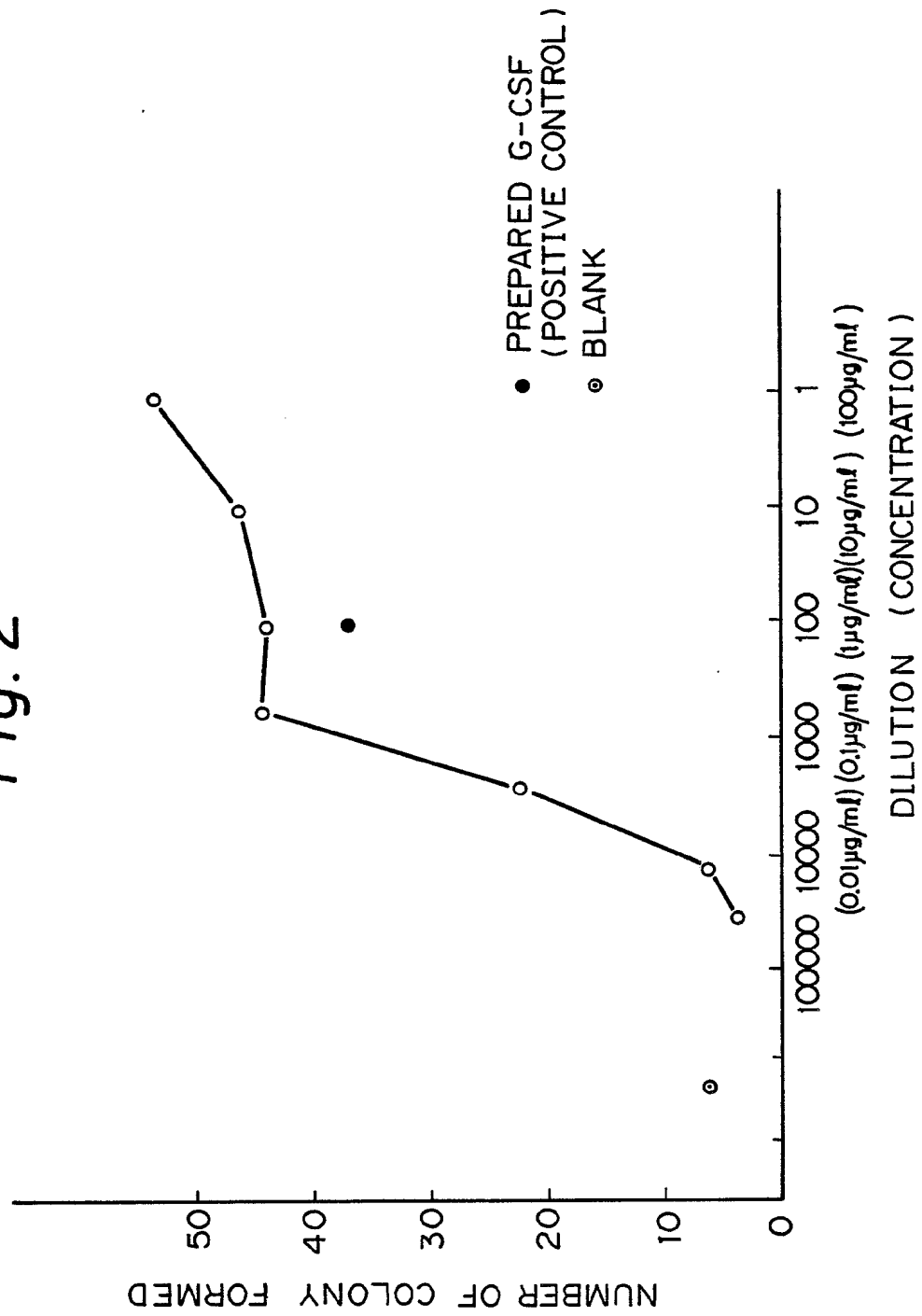


Fig. 3a

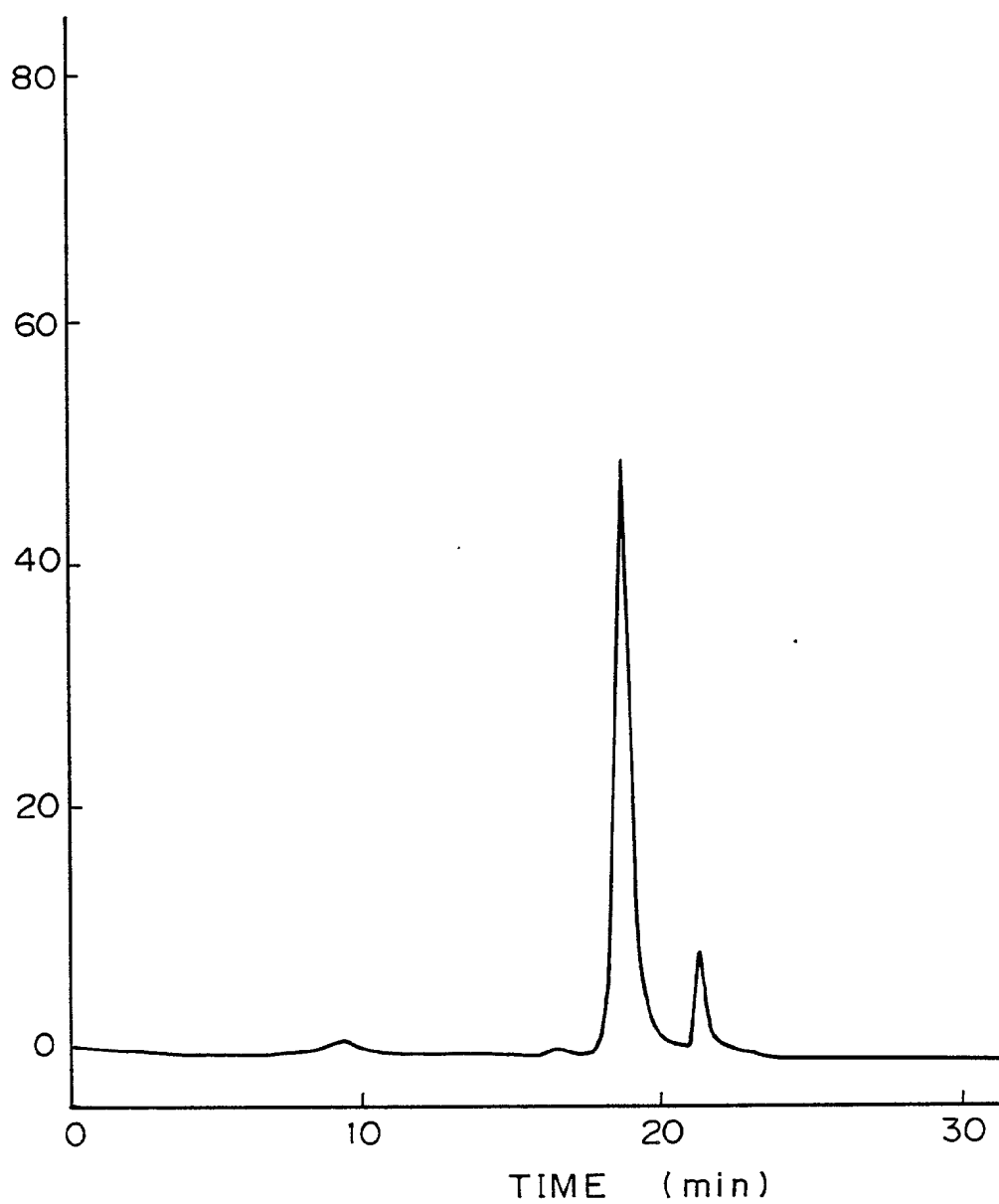


Fig. 3b

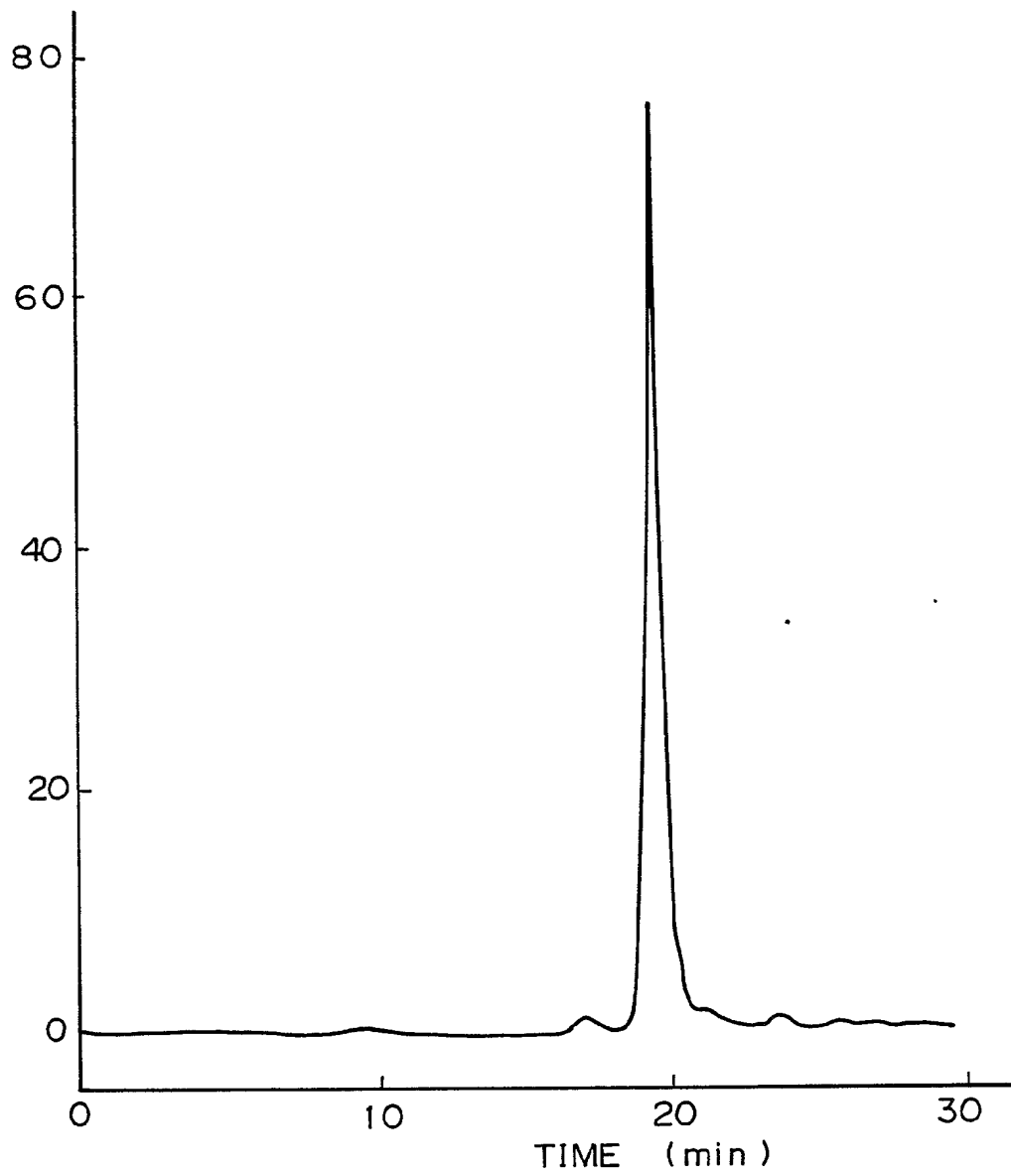


Fig. 4

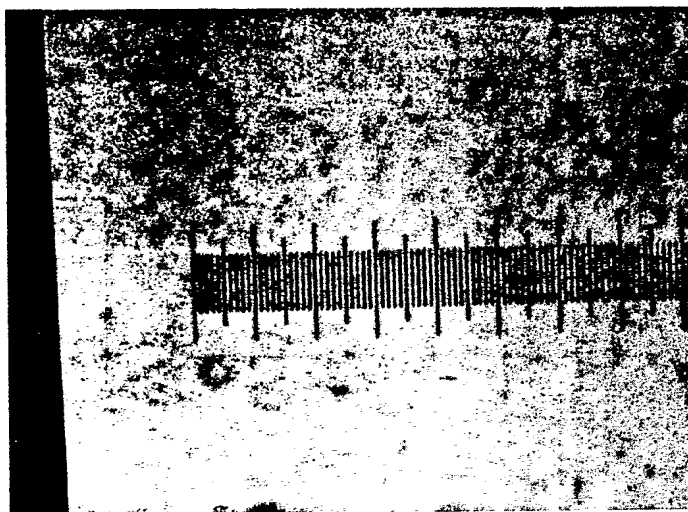


Fig. 5

